## The Crystal Structure of the 1:2-Complex of Ditoluenechromium and 7.7.8.8-Tetracyanoquinodimethane, $[(C_6H_5CH_3)_2Cr]^+(C_{12}H_4N_4)^-(C_{12}H_4N_4)$

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Summary The structure of [(C6H5CH3)2Cr]+ TCNQ-TCNQ<sup>0</sup> consists of stacks of ditoluenechromium cations and stacks of alternate TCNQ- and TCNQ<sup>0</sup>, with interplanar spacings of 3.29 Å.

EARLIER, we reported the crystal structure of the ionic 1:1-complex of ditoluenechromium and 7,7,8,8-tetracyanoquinodimethane, [(C<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>)<sub>2</sub>Cr]+ TCNQ<sup>-</sup> (I),<sup>1</sup> and we now present the results of an X-ray study on the 1:2-complex of ditoluenechromium and 7,7,8,8-tetracyanoquinodimethane [(C<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>)<sub>2</sub>Cr]<sup>+</sup> TCNQ<sup>-</sup> TCNQ<sup>0</sup> (II). A comparison between the structures of the two compounds is of importance, as they differ significantly in electrical conductivity.<sup>2</sup> For (I), the specific resistance measured on monocrystals is  $\rho = 2.5 \, 10^5 \, \text{ohm/cm.}$ , with activation energy  $E_{a} = 0.36$  ev; for (II),  $\rho = 0.5$  ohm/cm., at  $E_{a} =$ 0.08 ev.

Crystal data for (II): C38H24N8Cr, dark violet needles developed along the b axis, triclinic, space group P1,



FIGURE 1. The [100] projection of the [(C6H5CH3)2Cr]+ TCNQ-TCNQ<sup>0</sup> structure.

 $a = 8.25 \pm 0.02; \ b = 7.76 \pm 0.02; \ c = 13.77 \pm 0.03 \text{ Å};$  $\alpha = 94.7^{\circ} \pm 0.5^{\circ}; \ \beta = 92.3^{\circ} \pm 0.5^{\circ}; \ \gamma = 112.5^{\circ} \pm 0.5^{\circ};$  $U = 813.3 \text{ Å}^3$ ; M = 644.7; for  $Z = 1 D_c = 1.32 \text{ g./cm.}^3$ 

The intensities of ca. 1770 independent non-zero reflections (h0l-h6l and 0kl layers) were taken with a multiple-film equi-inclination Weissenberg goniometer with unfiltered Cu radiation, and estimated visually. A spot-size correction was applied.<sup>3</sup> Absorption corrections were deemed to be unnecessary ( $\mu = 33.6$  cm.<sup>-1</sup>).

If the space group is  $P\overline{1}$ , the lone  $[(C_6H_5CH_3)_2Cr]^+$  ion must occupy a centre of symmetry, while the two TCNQ molecules may lie either in two other centres of symmetry or else take a general position. A trial structure consistent with these assumptions was readily deduced by inspection of the three-dimensional Patterson synthesis. The structure (II) was refined by means of Fourier synthesis, followed by a full-matrix least-squares method with isotropic temperature factors to R = 0.13.

Isolated stacks of  $[(C_6H_5CH_3)_2Cr]^+$  cations and stacks of alternate TCNQ- and TCNQ<sup>o</sup> are essential features of the structure (II) (Figure 1). In the cation stack, one ion  $[(C_6H_5CH_3)_2Cr]^+$  falls within the period b. The cation of ditoluenechromium has a prismatic trans-configuration. The interplanar spacing of aromatic rings within a cation is  $3.34 \pm 0.04$  Å and for two adjacent cations in the stack it is  $3.\overline{68} \pm 0.04$  Å.  $(Cr-C)_{av} = 2.18 \pm 0.03$  Å;  $(C_{ar}-C_{ar})_{av}$  $= 1.40 \pm 0.04$  Å.

The molecules of TCNQ- and TCNQ<sup>0</sup> occupy two crystallographically independent centres of symmetry, but the accuracy of determination of bond lengths  $(\pm 0.03 \text{ Å})$ does not permit us to distinguish between TCNQ- and TCNQ<sup>0</sup>. The interplanar spacing of molecules TCNQ<sup>-</sup> and TCNQ<sup>0</sup> is  $3\cdot 29 \pm 0\cdot 03$  Å. The overlapping of neighbouring TCNQ<sup>-</sup> and TCNQ<sup>0</sup> in the stack is shown in Figure 2.



FIGURE 2. Overlapping of adjacent TCNQ<sup>-</sup> and TCNQ<sup>0</sup> in a stack.

If in the structure (I) each stack of TCNQ is surrounded by six stacks of cations as though making a barrier for interaction between TCNQ stacks, then the stacks of TCNQ in the structure (II) form layers parallel to the plane ab.

We thank Drs. E. B. Yagubskii, I. F. Shchegolev, and M. L. Khidekel for supplying the crystals and discussing the results.

(Received, September 15th, 1969; Com. 1396.)

 R. P. Shibaeva, L. O. Atovmyan, and L. P. Rozenberg, Chem. Comm., 1969, 649.
E. B. Yagubskii, M. L. Khidekel, G. F. Shchegolev, L. I. Buravov, B. G. Gribov, and M. K. Makova, Izvest. Akad. Nauk S.S.S.R., Ser. khim., 1968, 2124.

<sup>8</sup> D. C. Phillips, Acta Cryst., 1954, 7, 746; 1956, 9, 819.